

MFe₂O₄ (M= Mn, Co and Ni) nanoferrites: A simple solvothermal synthesis

S. Castro-García¹, S. Yáñez-Vilar¹, M. Sánchez-Andújar¹, C. Gómez-Aguirre¹, J. Mira², M. A. Señaris-Rodríguez¹

¹*Dept. Química Fundamental, Univ. da Coruña, A Zapateira s/n, 15071 A Coruña, España*

²*Dept. Física Aplicada, Univ. de Santiago de Compostela, 15781 Santiago de Compostela, España*

suqui@udc.es

During the past decades, the design, preparation and characterization of nanometric magnetic materials have been an important area of research that is attracting growing interest because of the potential applications that such materials have in ferrofluids, advanced magnetic materials and medical diagnostic [1].

In particular, the MFe₂O₄ (M= Mn, Fe, Co...) spinel ferrites are among the most important magnetic materials and have been widely used in electronic devices and for information storage [2]. A big variety of methods have been employed in the synthesis of these compounds, such as for example: sol-gel techniques, coprecipitation, mechanochemical processing, reverse micelles and microemulsions procedures, but the most cost-effective routes for large-production ferrite nanocrystals are the chemical precipitation and the solvothermal synthesis.

In this contribution, we report a simple preparation method to obtain MFe₂O₄ (M= Mn, Co and Ni) nanoparticles, that is an adaptation of the synthetic route described by N. Pinna et al. [3] to obtain Fe₃O₄ nanocrystals. This procedure involves a solvothermal reaction that is a valuable alternative to, for example, the known aqueous sol-gel process. And as it will be shown, the proposed method offers the following significant advantages for the preparation of ferrites: its simplicity, the high crystallinity of the obtained products at relatively low temperature (T~180 °C), the capability to control the crystal growth without the use of surfactants and its adequacy for the preparation of large amount of sample.

By this method, the synthesis of MFe₂O₄ (M= Mn, Co and Ni) is carried out under autogenous pressure inside a stainless-steel autoclave and an alcohol (hexanol or benzyl alcohol) is used both as solvent and ligand, so that it is not necessary to add a surfactant.

We will discuss the influence of the synthetic conditions used in the obtention of the highly crystalline nanoparticles (see Figure), and we will report a morphological, structural and magnetic characterization of the as-obtained nanoparticles.

Acknowledgments: The authors are grateful for the financial support from Consolider-Ingenio 2010 under project CSD2006-00012.

References:

- [1] F. Caruso, M. Spasova, A. Susha, M. Giersig, R. A. Caruso, *Chem. Mater.* **13** (2001) 109.
- [2] M. Schaefer, G. Dietzmann, H. Wriht, *J. Mag. Mag. Mater.* **101** (1991) 95.
- [3] N. Pinna, S. Grancharov, P. Beato, P. Bonville, M. Antonietti, M. Niederberger, *Chem. Mater.* **17** (2005) 3044.

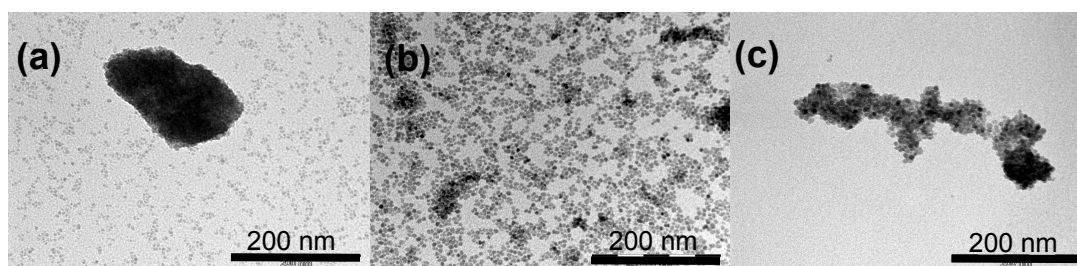


Figure. TEM micrographs of the (a) MnFe₂O₄, (b) CoFe₂O₄ and (c) NiFe₂O₄ nanoparticles synthesized in hexanol at 180 °C during 24 h.