## Nano scale surface and reaction mechanism study of (R)- and (S)-2-butanol over 100 surface alumina: Experimental vs. DFT

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Reactions of 2-butanol over alumina were investigated under various conditions. The mechanism of dehydration, dehydrogenation, ether formation, hydrogen-shift, and the conformational analysis of adsorbed (*R*)- and (*S*)-2-butanol over the (100) surface of alumina was computed by Density Functional Theory (DFT). Adsorption, dissociation energy, and Mulliken charges were calculated for (*R*)- and (*S*)-2-butanol adsorbed on the surface (100) alumina. Mulliken atomic charges predict that selected basic sites play a major role in elimination reactions. Computed conformational analysis indicates that the (*S*)-isomer forms a stronger bond with the surface than the (*R*)-isomer. Migration of proton of (*S*)-isomer (sc-ap conformer) to oxygen O<sub>c</sub> is an endothermic process (+9.187 kcal mol<sup>-1</sup>) with small activation energy (13.560 kcal mol<sup>-1</sup>). Transition state model, made of two layers of alumina is involved in anti elimination of the sc-sc conformer of (*S*)-2-butanol. E2 elimination with synclinal transition state is comparable with E2 antiperiplanar transition state (Figure 1).

The distance between basic sites of alumina and  $\beta$ , activation energy (E<sub>a</sub>), and enthalpy ( $\Delta H^o$ ) of the most stable conformers of adsorbed (R)- and (S)-2-butanol over the (100) surface were calculated to help shed light on the elimination reaction pathway. All basic oxygen sites of the surface at distances less than 4 Å from  $\beta$  hydrogens were considered as a site for elimination and were included in our study.

## **References:**

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Figure 1.

