SIZE TUNING AND OXYGEN PLASMA INDUCED PORE FORMATION ON SILICA NANOPARTICLES

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Silica nanoparticles (SNPs) occupy a prominent position in scientific research because of their easiness in preparation and enormous uses in various applications [1]. Porous silica nanoparticles have a promising role in various drug delivery applications [2, 3]. Significant research progress has been made in controlling and modifying the properties of mesoporous silica materials since its discovery [4-6]. In contrast to the earlier reports of tuning the particle size by adjusting atleast four different parameters [7, 8], here we are reporting about our attempt to control the nanoparticle size by making variations in a single parameter; the concentration of ammonia solution. Silica nanoparticles of size ranging from 5 nm to 250 nm have been successfully synthesized by controlling only the concentration of ammonia solution keeping all other parameters (concentration of TEOS, ethanol and water) constant [Fig 1]. Oxygen plasma was found to be a successful direct tool for generating pores on silica nanoparticles without the use of any structure directing agent and our method proved to be very easy and time saving one [Fig 2]. The nature and morphology of nanoparticles were investigated by scanning electron microscopy, transmission electron microscopy, dynamic light scattering, energy dispersive X-ray spectroscopy and Fourier transformed infrared spectroscopy.

References:

- [1] K.S. Rao et al., J. Colloid Interface Sci., 289 (2005) 125-131.
- [2] I.I. Slowing et al., Adv. Drug Del. Rev., **60** (2008) 1278-1288.
- [3] F. Torney et al., Nat. Nanotech., 2 (2007) 295-300.
- [4] G. Bogush et al., J. Non-Cryst. Solids, 104 (1988) 95-106.
- [5] T. Yokoi et al., J. Am. Chem. Soc., 128 (2006) 13664-13665.
- [6] J. Guo et al., J. Colloid Interface Sci., 326 (2008) 138-142.
- [7] S. K. Park et al., Colloids Surf. A., 197 (2002) 7-17.
- [8] Davies G-L et al., Chem. Phy. Lett., 468 (2009) 239-244.

Figures:

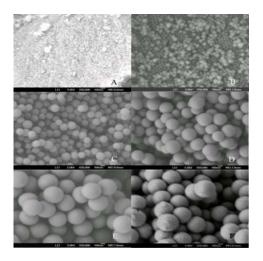


Fig 1: SEM images of silica nanoparticles at different concentrations of ammonia solution

A: Silica nanoparticles at 6% NH₃ solution (Average diameter: 9.16 nm)

B: Silica nanoparticles at 8% NH₃ solution (Average diameter: 67.81 nm)

C: Silica nanoparticles at 12% NH₃ solution (Average diameter: 137.66 nm)

D: Silica nanoparticles at 16% NH₃ solution (Average diameter: 216.62 nm)

E: Silica nanoparticles at 20% NH₃ solution (Average diameter: 242.07 nm)

F: Silica nanoparticles at 24% NH₃ solution (Average diameter: 257.03 nm)

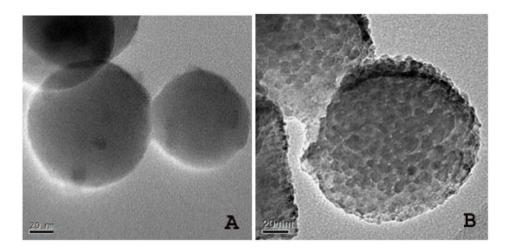


Fig 2: TEM image of silica nanoparticles with and without plasma treatment