

## Formation of enantioselective molecular structures on the PdGa:A(-1-1-1)Pd<sub>3</sub> surface

Stolz Samuel, Widmer Roland

EMPA, nanotech@surface, Überlandstrasse 191, 8600 Dübendorf, Switzerland  
[samuel.stolz@empa.ch](mailto:samuel.stolz@empa.ch)

### Abstract

Intermetallic PdGa has recently attracted considerable interest because of its high activity, selectivity, and stability in the catalytic semi-hydrogenation of ethyne [1], being an important step in the polyethylene production. Owing to its non-centrosymmetric bulk structure of the space group P2<sub>1</sub>3, PdGa exists in two enantiomeric forms A and B [2]. Investigating PdGa{111} surfaces, the stacking sequence in the [111] direction involves four nonequivalent atomic planes. As a consequence, the top and bottom surfaces are different: PdGa:A(111) is terminated by an atomic layer containing one isolated Pd atom per surface unit cell and accordingly this termination is denoted as Pd<sub>1</sub>, on the other side, PdGa:A(-1-1-1) reveals isolated Pd trimers and is thus denoted as Pd<sub>3</sub> [3].

To probe the chirality of the Pd<sub>1</sub> and Pd<sub>3</sub> surfaces, 9-Ethynylphenanthrene (9-EP) which is a prochiral molecule is adsorbed on both surfaces. For the Pd<sub>1</sub> surface a highly enantioselective adsorption with an enantiomeric excess of 98% is reported upon 9-EP evaporation at room temperature (RT) [4]. Adsorbing 9-EP at RT on Pd<sub>3</sub> yields in a 1:1 ratio of left and right oriented molecules, thus no enantiomeric excess is observed. However, post-annealing to 500 K of the 9-EP on Pd<sub>3</sub> results in the formation of new structures consisting of three 9-EP molecules in a propeller-like shape. 99.8% of the 9-EP molecules forming the propellers are of the same enantiomeric form, thus 99.3% of all propellers are homochiral with an enantiomeric excess of 98.6%.

Further analysis of these propellers by STM reveal a voltage dependent protrusion in the center of the propeller, which might be explained by a sp<sup>2</sup> hybridisation of the C=C-H bonds or by an electronic effect. Therefore, nc-AFM investigations with a CO functionalized tip were performed and will be compared to STM (see Fig. 1) and discussed.

### References

- [1] Marc Armbrüster, Kirill Kovnir, Malte Behrens, Detre Teschner, Yuri Grin and Robert Schlögl, *J. Am. Chem. Soc.*, **132** (2010) 14745-14747.
- [2] Dirk Rosenthal, Roland Widmer, Ronald Wagner, Peter Gille, Marc Armbrüster, Yuri Grin, Robert Schlögl, and Oliver Gröning, *Langmuir*, **28** (2012), 6848-6856
- [3] Jan Prinz, Roberto Gaspari, Carlo A. Pignedoli, Jochen Vogt, Peter Gille, Marc Armbrüster, Harald Brune, Oliver Gröning, Daniele Passerone and Roland Widmer, *Angew. Chem. Int. Ed.*, **51** (2012), 9339-9343.
- [4] Jan Prinz, Oliver Gröning, Harald Brune and Roland Widmer, *Angew. Chem.* **127** (2015), 3974-3978.

**Figure 1:** Comparison of 9-EP propellers on Pd<sub>3</sub> between STM and nc-AFM

